

Available Online at http://www.recentscientific.com

CODEN: IJRSFP (USA)

International Journal of Recent Scientific Research Vol. 8, Issue, 7, pp. 18756-18769, July, 2017 International Journal of Recent Scientific Research

DOI: 10.24327/IJRSR

Research Article

SYNTHESIS AND CHARECTERIZATION OF DIPYRIDAMOLE IMPURITIES BY SEQUENTIAL NUCLEOPHYLLIC SUBSTITUTION REACTION

Menaka, T* and Ramya Kuber, B

Sri Padmavati Mahila University, Tirupati, Andrapradesh, India

DOI: http://dx.doi.org/10.24327/ijrsr.2017.0807.0562

ARTICLE INFO

ABSTRACT

Article History: Received 05th April, 2017 Received in revised form 21st May, 2017 Accepted 06th June, 2017 Published online 28th July, 2017

Key Words: Dipyridamole, Impurities, Sequential nucleophilic substitution. Dipyridamole is used as drug for the treatment to inhibits blood clot formation and causes blood vessel dilation when given at high doses over a short time by inhibiting the phosphodiesterase enzymes that normally break down cAMP by increasing cellular cAMP levels and blocking the platelet aggregation response to ADP and cGMP. Presence of higher level of related substances or impurities may have harmful effect on body, hence needed to be identified, synthesised &characterised for safer use of the medicine. During process optimization of Dipyridamole drug, impurities were observed. These related substances or impurities were synthesised, characterized and proposed structures we reconfirmed by chemical synthesis. Dipyridamole impurities containing pyrimido-pyrimidine have been synthesized by the reaction of 2,4,6,8-tetrachloropyrimido[5,4-d] pyrimidine with sequential nucleophilic substitutions of piperidine, diethanolamine and ethanolamine in the pattern of C-4, C-8, C-2 and C-6 respectively³. Which have been characterized by using LCMS, ¹H NMR and HPLC analysis.

Copyright © **Menaka, T and Ramya Kuber, B, 2017**, this is an open-access article distributed under the terms of the Creative Commons Attribution License, which permits unrestricted use, distribution and reproduction in any medium, provided the original work is properly cited.

INTRODUCTION

Dipyridamole is used as drug for the treatment to inhibits blood clot formation. It is used to dilate blood vessels (Boehringer Ingelheim Pharmaceuticals, Inc. 2016) in people with peripheral arterial disease and coronary artery disease and it has been shown to lower pulmonary hypertension without significant drop of systemic blood pressure (Brown DG et al, 2015; Dixon BS et al 2009; Derendorf H et al, 2005). The safety of a drug product is not only dependent on the toxicological properties of the active drug substance or API (De Schryver EL et al, 2007; Sprigg N et al, 2008), but also on the impurities formed during the various chemical transformations. Therefore, identification, quantification, and control of impurities in the drug substance and drug product are important parts of drug development for obtaining marketing approval (S.J. Ingale et al, 2011; Brown D.G et al, 2015). As per the guidelines recommended by ICH, the acceptable level for a known orUnknown impurity or related substances is less than 0.2% in a drug substance. Tomeet the stringent regulatory requirements, it is more challenging for pharma industry to identify the impurities which are formed in very small quantities in a drug substance. Since most of the time it is very difficult to identify and control impurities within acceptable levels in the process, extra purification steps may then be necessary thereby making the process less competitive (V.S. Tegeli et al, 2011; B. Misralet al, 2015). The syntheses of impurities are not described in the literature which makes it even more difficult for the organic chemist who must then design a synthesis, which is time consuming. The development of a drug substance is incomplete without the identification of an impurity profile involved in the process. In our study we explored the identification, synthesis and characterization of impurities found in the preparation of dipyridamole. This study will be of immense help for the pharma industry to understand the potential impurities in dipyridamole synthesis and thereby obtain the pure compound (FA Attaby et al, 1995; SS Ghabrial et al, 1996; S Ranjit Pada et al, 2012). The object of this invention to provide novel synthetic route with high purity of dipyridamole impurities containing pyrimidopyrimidine nucleus, upon further study of the specification and appended claims, Pyrimidopyrimidines have been drawn as promising structural units in the field of medicinal chemistry. Pyrimido-pyrimidine derivatives exhibit various types of physiological activity and enter medicinal products and this determines the great attention that has been paid to the synthesis of new compounds of this series. The pyrimido-pyrimidine derivatives drew a lot of attention on various pharmacological activities because of their structural similarity to Purines, like anti-

Sri Padmavati Mahila University, Tirupati, Andrapradesh, India

microbial (SS Ghabrial *et al*, 1994; M. Suresh *et al*, 2010), antidiabetic, antioxidant (R Ghahremanzadeh *et al*, 2008) and antitumor (NA Devi *et al*, 1998; Osama Mohamed Ahmed *et al*, 2011; M Samir *et al*, 2011).

To this end we haveinvestigated how to control the substitution chemistry of 2, 4, 6, 8-tetrachloropyrimido [5, 4-d] pyrimidine. Besides enabling the synthesis of individual purine mimetics, a better knowledge of the chemistry of 2, 4, 6,8-tetrachloropyrimido [5, 4-d] pyrimidines would provide an opportunity for parallel or combinatorial syntheses of adverse array of novel compounds. 2, 4, 6, 8-tetrachloropyrimido [5, 4-d] pyrimidine is the classical precursor, the pattern of substitution is the typical outcome of two stepwise substitution reactions of 2, 4, 6, 8-tetrachloropyrimido [5, 4-d] pyrimidine [5, 4-d] pyrimidine, employing an excess of nucleophile in each step. This is because the rates of substitution greatly favour the C-4 and C-8 positions over C-2 and C-6. For example, treatment 2, 4, 6, 8-tetrachloropyrimido [5, 4-d] pyrimidine with an excess of piperidine at ambient temperature led to the formation of 4,8-bis(Piperidine)-2,6- dichloropyrimido [5,4-d] pyrimidine (*Yogesh D. Bommegowda et al, 2012*). However, no synthetic details have been reported. In this context, the present study describes identification, synthesis and characterization of dipyridamole impurities.

MATERIALS AND METHODS

Thin layer chromatography was used to monitor the completion of the reaction and homogeneity of the synthesised compound. Melting points were determined using a manual Buchi electro thermal apparatus (range $0-300^{\circ}$ C) in open capillary tubes and uncorrected.







Mass spectra were obtained using MS- Agilent technologies- 6130 Quadrupole LC/MS. ¹H-NMR spectra were recorded on Agilent varien 3.2 version,400 MHz in DMSO-d6 /CDCl3 with TMS as an internal standard, yields mentioned are from unoptimized reaction condition of isolated pure product. The homogeneity of the compounds was checked on silica gel coated plates, hexane, ethyl acetate and chloroform as the eluent and observed in UV lamp, iodine vapours or KMnO₄ spray as developing agents. All the synthesised compounds gave satisfactory elemental analysis.

RESULTS AND DISCUSSION

Scheme-1: Synthesis of impurity-1 or impurity A as per EP monograph



Step-1; Synthesis of 4, 8-bis (piperidine)-2, 6- dichloropyrimido[5,4-d] pyrimidine.



To 2, 4, 6, 8-tetrachloropyrimido [5, 4-d] pyrimidine (2.0g, 7.40 mmol) in dry THF (50–60ml) stirred under nitrogen atmosphere at 25°C, was added an appropriate piperidine (3.23g, 3.24ml, 29.60mmol), 2.40 equiv. potassium carbonate. The resulting suspension was agitated for 20 min. Addition of water (80 ml) precipitated the product. The solid was filtered off, washed with water and dried. The resulting product was recrystallised from hot ethyl acetate gave the title compound as fine yellow solid. (2.20g, 74%); mp 202-

206°c, LC-MS: m/z 367.25 [M+1]. 1H NMR (CDCl3): δ 1.58-1.62 (12H, m,-CH2-); δ 2.90-3.16 (8H, m,-CH2-). Anal. Calcd for C₁₆H₂₀C₁₂N₆: C, 52.32: H, 5.49: N, 22.88. Found: C, 52.00: H, 5.72: N, 22.62.

Step-2; Synthesis of 2-chloro-4,6,8-tri(piperidin-1-yl)pyrimido[5,4-d]pyrimidine



The 4, 8- bis Piperidine-2, 6-dichloropyrimido [5, 4-d] pyrimidine (0.20 g, 0.24 mmol) dissolved in dry THF under nitrogen, piperidine was added slowly dropwise, the reaction mixture was stirred for 1 hr at room temperature, water (10ml) was added and stirred for 30 minutes, solid was precipitated washed with excess of water to get 99% of solid crystals (0.25 g,). The product was confirmed by LC-MS: m/z 416.2.

Step-3; Synthesis of 2,2'-[[4,6,8-Tri(piperidin-1-yl)pyrimido[5,4-d]pyrimidin-2-yl]nitrilo] diethanol



To the step-2, product: 2-chloro-4,6,8-tri(piperidin-1-yl) pyrimido[5,4-d] pyrimidine (0.20 g, 0.24 mmol) dissolved in DMSO, added diethanolamine dropwise at RT, temperature raised to 120° C for 3-4 hours, reaction was monitored by TLC, reaction mixture was cooled to 50°C, acetone and water 50:50 was added, stirred for 1 hr at room temperature, solid was precipitated, filtered off the reaction mixture, the solid was washed with water followed by hexane and ethyl acetate 60:40 twice to get yellow solid product. (0.2g 99%); mp 202-206°c, LC-MS: m/z 485.4 [M+1]. 1H NMR (DMSO- d6): δ 1.58-1.80 (30H, m,-CH2-); δ 3.30-4.04 (8H, m,-CH2-). Anal. Calcd for C₂₅H₄₀N₈O₂: C, 61.96; H, 8.32; N, 23.12; O, 6.60 Found: C, 62.00: H, 8.50: N, 24.05.O, 6.45.

Scheme-2: Synthesis of impurity-2 or impurity B as per EP monograph



Step-1: Synthesis of 8-piperidine-2, 4, 6- trichloropyrimido [5, 4-d] pyrimidine



To 2,4,6,8,tetra chloro pyrimido[5,4-d]pyrimidine(1.0 g,0.37 mmol) and potassium carbonate (1.0g, 0.74 mmol) in THF (60 ml), stirred under nitrogen at -78°c, piperidinewas added (0.40 g, 0.42 mmol)in THF (60 ml) drop wise from a syringe at a rate of 1 ml per minute. A cloudy yellow solution was obtained and it was allowed to warm up for 10 min. addition of an excess of water gave fine yellow precipitate which was extracted with ethyl acetate. After drying of organic extract over sodium sulphate, removal of the solvent gave a yellow cake, triturate with diethyl ether, filtration and dry under vacuum to give the yellow coloured solid, which was used without further purification (0.90g, 78%). The product formation was confirmed by TLC and LCMS analysis.

Step-2: Synthesis of 6-chloro, 2,2',2'',2'''-[[8-(Piperidin-1-yl) pyrimido[5,4-d] pyrimidine-2,4-diiyl] diinitrilo]hexaethanol.



To the step-1 product, 8-piperidine-2, 4, 6- trichloropyrimido [5, 4-d] pyrimidine. (0.5 g,0.50mmol) dissolved in DMSO at room temperature, dropwise added diethanolamine (0.5 g, 3.0mmol) and heated to 120°C for 1 hr.the reaction mixture was cooled to 50°C added 50ml of water and acetone. The formed solid was washed with hexane 3-4 times followed by ethyl acetate, which was used without further purification (0.60g, 80%).

Step-3: Synthesis of 2,2',2'',2''',2''''-[[8-(piperidin-1-yl) pyrimido[5,4-d] pyrimidine-2,4,6-triyl]trinitrilo]hexaethanol



Impurity-2

To the step-2 product,6-chloro, 2,2',2",2"'-[[8-(Piperidin-1-yl) pyrimido[5,4-d] pyrimidine-2,4-diiyl] diinitrilo] hexaethanol (0.5 g,0.45mmol) dissolved in DMSO at room temperature, dropwise added diethanolamine (0.2 g,1.5mmol) and heated to 120°C for 12 hrs. the reaction mixture was cooled to 50°C added 50ml of water stirred for 50 min at 50°C. Reaction mixture was washed with hexane 3-4 times followed by ethyl acetate twice to get yellow solid, (0.4g 95%); mp 202-206°c, LC-MS: m/z 525.3 [M+1]. 1H NMR (DMSO-d6): δ 1.59-2.49 (8H, m, -CH2-); δ 3.56-3.70 (28H, m,-CH2-). Anal. Calcd for C₂₃H₄₀N₈O₆: C, 52.66; H, 7.69; N, 21.36; O, 18.30 Found: C, 52.50: H, 7.98: N, 21.63.O, 18.80.

Scheme-3: Synthesis of impurity-3 or impurity C as per EP monograph



Step-1; Synthesis of 4, 8-bis (piperidine)-2, 6- dichloropyrimido[5,4-d] pyrimidine



To 4,8-bis(Piperidine)-2,6- dichloropyrimido [5,4-d] pyrimidine (2.0g, 7.40 mmol) in dry THF (50–60ml) stirred under N₂ at 25°C, was added an appropriate piperidine (3.23g,3.24ml,29.60mmol),2.40 equiv. potassium carbonate. The resulting suspension was agitated for 20 min. Addition of water (80 ml) precipitated the product. The solid was filtered off, washed with water and dried. The resulting product was recrystallised from hot ethyl acetate gave the title compound as fine yellow solid. (2.20g,74%); mp 202-206°c, LC-MS: m/z 367.25 [M+1]. 1H NMR (CDCl3): δ 1.58-1.62 (12H, m,-CH2-); δ 2.90-3.16 (8H, m,-CH2-). Anal. Calcd for C₁₆H₂₀C₁₂N₆: C, 52.32: H, 5.49: N, 22.88. Found: C, 52.00: H, 5.72: N, 22.62.

Step-2; Synthesis of 2,2'-[[2-Chloro-4,8-di(piperidin-1-yl) pyrimido[5,4-d]pyrimidin-6-yl]nitrilo] diethanol.



To a stirred solution of step-1 product, 4, 8-bis (piperidine)-2, 6- dichloropyrimido[5,4-d] pyrimidine (0.30 g, 0.88 mmol) in dry THF (12ml) under nitrogen at 0°c was added appropriate dihydroxy ethanolamine (>2.0 mol equiv. 33%,0.8 ml) in THF (12 ml) over a 10min period. Precipitate formed and suspension was left to stir for 10-30min. reaction was monitored by TLC, reaction mixture was cooled and 10 volumes of water was added to precipitate the product, filtered off, washed with hexane to get yellow solid. (2.20g,74%); mp 202-206°c, LC-MS: m/z 436.2 [M+1]. 1H NMR (DMSO-d6): δ 1.58-1.64 (20H, m, -CH2-); δ 3.6 (8H, m,-CH2-). Anal. Calcd for C₂₀H₃₀ClN₇O₂: C, 55.10; H, 6.94; Cl, 8.13; N, 22.49; O, 7.34 Found: C, 55.50: H, 7.25: Cl, 7.98; N, 22.56; O, 7.20.





Impurity-4

Step-1; Synthesis of 4, 8-bis (piperidine)-2, 6- dichloropyrimido[5,4-d] pyrimidine.



To 2, 4, 6, 8-tetrachloropyrimido [5, 4-d] pyrimidine (2.0g, 7.40 mmol) in dry THF (50–60ml) stirred under nitrogen atmosphere at 25°C, was added an appropriate piperidine (3.23g, 3.24ml, 29.60mmol), 2.40 equiv. potassium carbonate. The resulting suspension was agitated for 20 min. Addition of water (80 ml) precipitated the product. The solid was filtered off, washed with water and dried. The resulting product was recrystallised from hot ethyl acetate gave the title compound as fine yellow solid. (2.20g, 74%); mp 202-206°c, LC-MS: m/z 367.25 [M+1]. 1H NMR (CDCl3): δ 1.58-1.62 (12H, m, -CH2-); δ 2.90-3.16 (8H, m, -CH2-). Anal. Calcd for C₁₆H₂₀C₁₂N₆: C, 52.32: H, 5.49: N, 22.88. Found: C, 52.00: H, 5.72: N, 22.62.

Step-2; Synthesis of 2,2'-[[2-Chloro-4,8-di(piperidin-1-yl) pyrimido[5,4-d] pyrimidin-6-yl] nitrilo] diethanol



To a stirred solution of step-1 product, 4, 8-bis (piperidine)-2, 6- dichloropyrimido[5,4-d] pyrimidine (0.30 g, 0.88 mmol) in dry THF (12ml) under nitrogen at 0°c was added appropriate dihydroxy ethanolamine (>2.0 mol equiv. 33%,0.8 ml) in THF (12 ml) over a 10min period. Precipitate formed and suspension was left to stir for 10-30min. reaction was monitored by TLC, reaction mixture was cooled and 10 volumes of water was added to precipitate the product, filtered off, washed with hexane to get yellow solid. (2.20g,74%); mp 202-206°c, LC-MS: m/z 436.2 [M+1]. 1H NMR (DMSO-d6): δ 1.58-1.64 (20H, m, -CH2-); δ 3.6 (8H, m, -CH2-). Anal. Calcd for C₂₀H₃₀ClN₇O₂: C, 55.10; H, 6.94; Cl, 8.13; N, 22.49; O, 7.34 Found: C, 55.50: H, 7.25: Cl, 7.98; N, 22.56; O, 7.20.

Step-3; Synthesis of 2,2-[[2-(2-Hydroxyethyl) amino-4,8-di(piperidin-1-yl)pyrimido[5,4-d]pyrimidin-6-yl]nitrilo]diethanol



Impurity-4

To the appropriate step-4 (0.20 g, 0.63 mmol) in dry THF (5–12 ml) under nitrogen was added ethanolamine (5 ml). The resultant yellow solution was heated at 65°C for 2–4 h. The suspension was cooled and water (50 ml) was added, precipitating the product. The resultant mixture was filtered, to give an off-white solid which was retained. The filtrate was extracted with ethyl acetate and the organic layer was washed with water, dried (Na2SO4), filtered and recrystalised from ethyl acetate, gave a title compound as a white solid (0.15g, 78%). mp 202-206°c, LC-MS: m/z 461.3[M+1]. 1H NMR (DMSO-d6): δ 1.59 (12H, m, -CH2-); δ 3.5-3.58 (8H, m, -CH2-); δ 4.0-4.6 (12H, m, -CH2-); δ 4.5 (1H, s-NH-). Anal. Calcd for C₂₂H₃₆N₈O₃: C, 57.37; H, 7.88; N, 24.33; O, 10.42 Found: C, 56.98; H, 7.65; N, 24.88; O, 9.98.





To 2,4,6,8, tetra chloro pyrimido[5,4-d] pyrimidine (1.0 g,0.37 mmol) and potassium carbonate (1.0g, 0.74 mmol) in THF (60 ml), stirred under nitrogen at -78°c, piperidine was added (0.40 g, 0.42 mmol) in THF(60 ml) drop wise from a syringe at a rate of 1 ml per minute. a cloudy yellow solution was obtained and it was allowed to warm up for 10 min. addition of an excess of water gave fine yellow precipitate which was extracted with ethyl acetate. After drying of organic extract over sodium sulphate, removal of the solvent gave a yellow cake, triturate with diethyl ether, filtration and dry under vacuum to give the yellow coloured solid, which wasused without further purification (0.90g, 78%). The product formation was confirmed by TLC and LCMS analysis.

Step-2: Synthesis of 6-chloro,2,4-di [di(2-hydroxyethyl) amino]-8-di(piperidin-1-yl)-pyrimido [5,4-d] pyrimidine.



To the step-1 product, 8-piperidine-2, 4, 6- trichloropyrimido [5, 4-d] pyrimidine (0.5 g, 0.50mmol) dissolved in DMSO at room temperature, dropwise added diethanolamine (0.5 g, 3.0mmol) and heated to 120° C for 1 hr. the reaction mixture was cooled to 50° C added 50ml of water and acetone. The formed solid was washed with hexane 3-4 times followed by ethyl acetate, which was used without further purification (0.60g, 80%).

Step-3: Synthesis of 2,4-Di[di(2-hydroxyethyl)amino]-6,8-di(piperidin-1-yl)-pyrimido[5,4-d] pyrimidine



Impurity-5

To the step-2,product: Synthesis of 6-chloro,2,4-di [di(2-hydroxyethyl) amino]-8-di(piperidin-1-yl)-pyrimido [5,4-d] pyrimidine (0.20 g, 0.24 mmol) dissolved in DMSO, added diethanolamine dropwise at RT, temperature raised to 120° C for 12 hours, reaction was monitored by TLC, reaction mixture was cooled to 50°C, acetone and water 50:50 was added, stirred for 1 hr at room temperature, solid was precipitated, filtered off the reaction mixture, the solid was washed with water followed by hexane and ethyl acetate 60:40 twice to get yellow solid product. (0.2g 99%); mp 202-206°c, LC-MS: m/z 505.62 [M+1]. 1H NMR (DMSO- d6): δ 1.48-1.60 (12H, m, -CH2-); δ 3.50-3.68 (16H, m, -CH2-); δ 4.0-4.7 (12H, m, -CH2-). Anal. Calcd for C₂₄H₄₀N₈O₄: C, 57.12; H, 7.99; N, 22.21; O, 12.68 Found: C, 56.84; H, 7.45; N, 23.50; O, 12.77.



Step-1; Synthesis of 4, 8-bis (piperidine)-2, 6- dichloropyrimido[5,4-d] pyrimidine



To 2, 4, 6, 8-tetrachloropyrimido [5, 4-d] pyrimidine (2.0g, 7.40 mmol) in dry THF (50–60ml) stirred under nitrogen atmosphere at 25°C, was added an appropriate piperidine (3.23g, 3.24ml, 29.60mmol), 2.40 equiv. potassium carbonate. The resulting suspension was agitated for 20 min. Addition of water (80 ml) precipitated the product. The solid was filtered off, washed with water and dried. The resulting product was recrystallised from hot ethyl acetate gave the title compound as fine yellow solid. (2.20g, 74%); mp 202-206°c, LC-MS: m/z 367.25 [M+1]. 1H NMR (CDCl3): δ 1.58-1.62 (12H, m, -CH2-); δ 2.90-3.16 (8H, m, -CH2-). Anal. Calcd for C₁₆H₂₀C₁₂N₆: C, 52.32: H, 5.49: N, 22.88. Found: C, 52.00: H, 5.72: N, 22.62.

Step-2: synthesis of 2-(2,6-dichloro-8-(piperidin-1-yl) pyrimido[5,4-d] pyrimidin-8-ylamino) ethanol.



To the step-1 product, 4-piperidine-2, 6, 8- trichloropyrimido [5, 4-d] pyrimidine. (0.5g, 0.50 mmol) dissolved in DMSO at room temperature, dropwise added ethanolamine (0.3g, 1.5mmol) and heated to 120°C for 1 hr. the reaction mixture was cooled to 50°C added 50ml of water and acetone. The formed solid was washed with hexane 3-4 times followed by ethyl acetate, which was used without further purification (0.48g, 80%).

Step-3: synthesis of 2,6-di[di(2-hydroxyethyl) amino]-4-(2-hydroxyethyl)amino-8-(piperidin-1-yl)-pyrimido[5,4-d] pyrimidine.



Impurity-6

To the step-2 product, 2-(2,6-dichloro-4-(piperidin-1-yl) pyrimido[5,4-d] pyrimidin-8-ylamino) ethanol. (0.25g,0.40mmol) dissolved in DMSO at room temperature, dropwise added diethanolamine (0.3g, 3.5mmol) and heated to 120°C for 12 hrs. the reaction mixture was cooled to 50°C added 50ml of water stirred for 50 min at 50°C. Reaction mixture was washed with hexane 3-4 times followed by ethyl acetate twice to get yellow solid, (0.4g 95%); mp 202-206°c, LC-MS: m/z 481.6 [M+1]. 1H NMR (DMSO-d6): δ 1.58-1.65 (6H, m, -CH2-); δ 3.40-3.62 (14H, m, -CH2-); δ 4.62-4.75 (10H, m, -CH2-); δ 4.14(1H, s, -NH-). Anal. Calcd for C₂₁H₃₆N₈O₅: C, 52.49; H, 7.55; N, 23.32; O, 16.65 Found: C, 52.32; H, 7.05; N, 20.82; O, 16.86.

Scheme-6: Synthesis of 4, 8-bis (piperidine)-2, 6- dichloropyrimido[5,4-d] pyrimidine.



To 2,4,6,8-tetrachloropyrimido[5,4-d] pyrimidine (2.0g, 7.40 mmol) in dry THF (50–60ml) stirred under N₂ at 25°C, was added an appropriate piperidine (3.23g,3.24ml,29.60mmol),2.40 equiv. potassium carbonate. The resulting suspension was agitated for 20 min. Addition of water (80 ml) precipitated the product. The solid was filtered off, washed with water and dried. The resulting product was recrystallised from hot ethyl acetate gave the title compound as fine yellow solid. (2.20g,74%); mp 202-206°c, LC-MS:

m/z 367.25 [M+1]. 1H NMR (CDCl3): δ 1.58-1.62 (12H, m,-CH2-); δ 2.90-3.16 (8H, m,-CH2-). Anal. Calcd for C₁₆H₂₀C₁₂N₆: C, 52.32: H, 5.49: N, 22.88. Found: C, 52.00: H, 5.72: N, 22.62.

Acknowledgement

Authors are thank fulto PADM Laboratories&Mallige college of pharmacy, Bangalore, for providinglaboratory and Sophisticated Instrumentation Facility.

CONCLUSION

For the better understanding of the synthetic pathway of an active pharmaceutical ingredient (API) it is necessary to identify all the impurities formed/anticipated. In this regard, we have synthesized and characterized different potential process-related impurities of dipyridamole.

References

boehringer ingelheim pharmaceuticals, inc. december 2016.

- brown d.g., wilkerson e.c., and love w.e. 2015. a review of traditional and novel oral anticoagulant and antiplatelet therapy for dermatologists and dermatologic surgeons, journal of the american academy of dermatology, 72, 524-534.
- Dixon B.S., Beck G.J., and Vazquez M.A. 2009. Effect of dipyridamole plus aspirin on hemodialysis graft patency. N Engl J Med. 360, 2191-2201.
- Derendorf H., VanderMaelen C.P., and Brickl R.S., MacGregor T.R., Eisert W. 2005. Dipyridamole bioavailability in subjects with reduced gastric acidity. *J Clin Pharmacol*, 845-850.
- De Schryver E., L Algra A., and van Gijn J. Dipyridamole for preventing stroke and other vascular events in patients with vascular disease. Cochrane Database of Systematic Reviews. 2007.
- Sprigg N, Gray L.J., England T., Berger J.S., A randomized controlled trial of triple antiplatelet therapy (aspirin, clopidogrel and dipyridamole) in the secondary prevention of stroke: safety, tolerability and feasibility. 2008.
- Brown D.G., Wilkerson E.C., and Love W.E. 2015. Journal of the American Academy of Dermatology, 72, 524-34.
- S.J. Ingale, C.M., Sahu, R.T., Paliwal, S., Vaidya., and A.K. Singhai., *International journal of Pharmacy & Life Sciences*, 2, 955-962.
- V.S., Tegeli, G.B., Gajeli, G.K., Chougule, Y.S., Thorat, U.S., Shivsharan., S.T., Kumbhar, 2011. International Journal of Drug Form. & Res., 2(4): 174-195.
- B. Misra1, A. Thakur, P.P., Mahata1. 2015. Int. J. of Pharm. Chem., 5 (7): 232-239.
- F.A., Attaby, S.M., Eldin; M.A., Razik. 1995. Phosphorus Sulfur., 21, 106.
- S.S., Ghabrial, S.M., Eldin, 1996. Egypt J. Pharm Sci., 37, 375.
- S. Ranjit Pada, N. Ram Nandaniya, K. Haresh, H. RamViresh, Shah. 2012. Journal of Chem. Pharm. Res., 2012, 4(7): 3557-3561. SS Ghabrial; M.Y Zaki; SM Eldin, *Indian J. Chem.*, 1994, 33B, 855.
- M. Suresh, P. Lavanya, K. Vasu, D. Sudhakar; C Venkata Rao, 2010. Journal of Chem. Pharm. Res., 2(2): 82-89.
- R. Ghahremanzadeh, S.C., Azimi, N. Gholami, A. Bazgir. 2008. Chem Pharm Bull. 56(11): 1617-20.
- N.A., Devi, C.K., Khuman, R. Singh, L.W. Singh. 1998. Indian Journal of Heterocyclic chemistry. 7(3): 193-196.
- Osama Mohamed Ahmed, Ahmed M Hussein, 2011. Sci Pharm., 79(3): 429-447.

M. Samir, E. Moghazy, A. Diaa, Ibrahim, M. Nagwa, Abdelgawad, A.H., Nahla, Farag Ahmad, S., 2011. *Sci Pharm*, 79, 429-447. Yogesh D. Bommegowda, Dinesh B. Shenoy, Menaka T, Chethan B P, Nagaraja G,

Fazil Baig, Sandeep K N and Venkanna Babu. 2012. Journal of Chemical and Pharmaceutical Research, 4(11): 4888-4893.

How to cite this article:

Menaka, T and Ramya Kuber, B.2017, Synthesis and Charecterization of Dipyridamole Impurities By Sequential Nucleophyllic Substitution Reaction. *Int J Recent Sci Res.* 8(7), pp. 18756-18769. DOI: http://dx.doi.org/10.24327/ijrsr.2017.0807.0562
